## Some More Examples of Metric Conversions

Here are some more examples. They follow the same procedure as the problem presented above. Some problems with **mass** and **volume** are also shown. They are worked the same way as the length problems.

1. Convert 3 kilometers (km) into millimeters (mm).

2. How many dekagrams (dag) are in 424 milligrams (mg)?

| Table # 2 C Metric Conversion Table For Mass |       |       |        |                                     |        |         |                    |                    |                     |
|--|-------|-------|--------|-------------------------------------|--------|---------|--------------------|--------------------|---------------------|
| <u>Number of Grams</u> in each unit          |       |       | 1 Unit | <u>Number of units</u> in each Gram |        |         |                    |                    |                     |
| 1000 g                                       | 100 g | 10 g  | 1 g    | 10 dg                               | 100 cg | 1000 mg | 10 <sup>6</sup> μg | 10 <sup>9</sup> ng | 10 <sup>12</sup> pg |
| 1 kg   | 1 hg  | 1 dag | 1 g    | 1 g                                 | 1 g    | 1 g     | 1 g                | 1 g                | 1 g                 |

Notice that in this problem, the ratio 1000 mg was flipped upside down so that **Amg@**will cancel out. The ratio 1 g for dekagrams was also flipped for the same reason.

## 3. What is the volume in hL of 758 ml of alcohol?

$$758 - \frac{\text{ml}}{\text{ml}} \times \frac{1 + \frac{1}{\text{L}}}{1000 - \frac{\text{ml}}{\text{ml}}} \times \frac{1 + \frac{1}{\text{hL}}}{100 - \frac{1}{\text{L}}} = .00758 \text{ hL}$$

The following page of practice problems can be printed. The answers for the problems are also given so that you can make a quick determination whether or not you are doing them correctly.

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## Assignment #1 is worksheet that follows these practice problems. Check the Assignment Schedule page.

Also there are several web sites on the Internet that have explanations and examples of problems. You may find these Web Sites helpful. Sometimes a different way of explaining an idea helps a student understand the concept better.

You will find these links in the Activities Page of the Chem One Web Site.